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Electrical conductivity of the lower-mantle ferropericlase across the electronic spin transition

Jung-Fu Lin,[1] Samuel T. Weir,[1] Damon D. Jackson,[1] William J. Evans,[1] Yogesh K. Vohra,[2] Wei Qiu,[2] and Choong-Shik Yoo[1]

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[1] Electrical conductivity of the lower-mantle ferropericlase-$(Mg_{0.75},Fe_{0.25})O$ has been studied using designer diamond anvils to pressures over one megabar and temperatures up to 500 K. The electrical conductivity of $(Mg_{0.75},Fe_{0.25})O$ gradually rises by an order of magnitude up to 50 GPa but decreases by a factor of approximately three between 50 to 70 GPa. This decrease in the electrical conductivity is attributed to the isosymmetric high-spin to low-spin transition of iron in ferropericlase. That is, the electronic spin transition of iron results in a decrease in the mobility and/or density of the charge transfer carriers in the low-spin ferropericlase. The activation energy of the low-spin ferropericlase is 0.27 eV at 101 GPa, consistent with the small polaron conduction (electronic hopping, charge transfer). Our results indicate that low-spin ferropericlase exhibits lower electrical conductivity than high-spin ferropericlase, which needs to be considered in future geomagnetic models for the lower mantle. **Citation:** Lin, J.-F., S. T. Weir, D. D. Jackson, W. J. Evans, Y. K. Vohra, W. Qiu, and C.-S. Yoo (2007), Electrical conductivity of the lower-mantle ferropericlase across the electronic spin transition, *Geophys. Res. Lett.*, *34*, L16305, doi:10.1029/2007GL030523.

1. Introduction

[2] Knowledge of the electrical conductivity of the mineral assemblage in the Earth's lower mantle, mainly ferropericlase [(Mg,Fe)O] and silicate perovskite [Al-(Mg,Fe)SiO_3], is essential to understanding the propagation of the geomagnetic signals to the Earth's surface, the nature of the core-mantle coupling [e.g., *Olsen*, 1999; *Buffet*, 1992, 1996; *Holme*, 1998a, 1998b], and the subtle chemistry of the deep Earth [e.g., *Dobson and Brodholt*, 2000a, 2000b; *Xu et al.*, 2000]. Although silicate perovskite is believed to be volumetrically the dominant phase in the lower mantle, it has been demonstrated that the electrical conductivity profile of the lower mantle can be modeled with a network of ferropericlase in an insulating matrix of silicate perovskite [*Wood and Nell*, 1991].

[3] The electrical conductivity of ferropericlase was previously measured up to 40 GPa and temperatures as high as ~3000 K [e.g., *Mao*, 1973; *Peyronneau and Poirier*, 1989; *Li and Jeanloz*, 1990; *Dobson et al.*, 1997, *Dobson and Brodholt*, 2000a]. These studies showed that the electrical conductivity of ferropericlase is very sensitive to the iron content, ferrous to ferric iron ratio, and point defects. However, the electrical conductivity of the low-spin ferropericlase has not been measured and the potential effect of the recently observed pressure-induced electronic spin-pairing transition of iron on the electrical conductivity of ferropericlase is still unknown [e.g., *Badro et al.*, 2003; *Lin et al.*, 2005, 2006a, 2006b, 2007; *Speziale et al.*, 2005; *Goncharov et al.*, 2006; *Persson et al.*, 2006; *Tsuchiya et al.*, 2006; *Keppler et al.*, 2007].

[4] Ferropericlase is a solid solution between periclase (MgO), a wide band gap insulator, and wüstite (FeO), a classical Mott insulator and an important member of the highly correlated transition metal monoxide (TMO) group [*Mott*, 1990; *Cohen et al.*, 1997]. The Mott insulator-metal transition results from the closure of the Mott-Hubbard *d-d* band gap or of the charge-transfer *p-d* gap, and has been theoretically and/or experimentally reported to occur in transition metal oxides such as FeO [*Cohen et al.*, 1997; *Knittle and Jeanloz*, 1986], MnO [*Patterson et al.*, 2004; *Yoo et al.*, 2005], and Fe_2O_3 [*Pasternak et al.*, 1999]. In this regard, it is thus interesting to understand the effect of the electronic spin transition on the electrical conductivity of ferropericlase with iron concentration that is above the percolation threshold of 12% for the face-centered cubic (fcc) lattice, in which Fe^{2+} atoms may form infinitely connected percolation path through the whole structure [*Lorenz and Ziff*, 1998; *Lin et al.*, 2006a].

[5] Abnormal effects of the electronic spin transition of iron on the volume, incompressibility, sound velocities, and optical absorption spectra of ferropericlase have been observed under high pressures and room temperatures; a decrease in volume and radiative thermal conductivity and an increase or a softening in incompressibility and sound velocities are reported to occur at high pressures and room temperature [*Lin et al.*, 2005, 2006b; *Speziale et al.*, 2005; *Goncharov et al.*, 2006; *Keppler et al.*, 2007]. These new results are changing our view of the physical and chemical states of the lower mantle. Here we have measured the electrical conductivity of ferropericlase-$(Mg_{0.75},Fe_{0.25})O$ in the diamond anvil cell (DAC) using designer diamond anvils to pressures across the spin-pairing transition and exceeding one megabar. We have also measured the temperature effect on the electrical conductivity of the low-spin ferropericlase and derived its activation energy.

2. Experiments

[6] Polycrystalline $(Mg_{0.75},Fe_{0.25})O$ was synthesized under a controlled oxygen fugacity near the iron-wüstite

[1]Lawrence Livermore National Laboratory, Livermore, California, USA.
[2]Department of Physics, University of Alabama at Birmingham, Birmingham, Alabama, USA.

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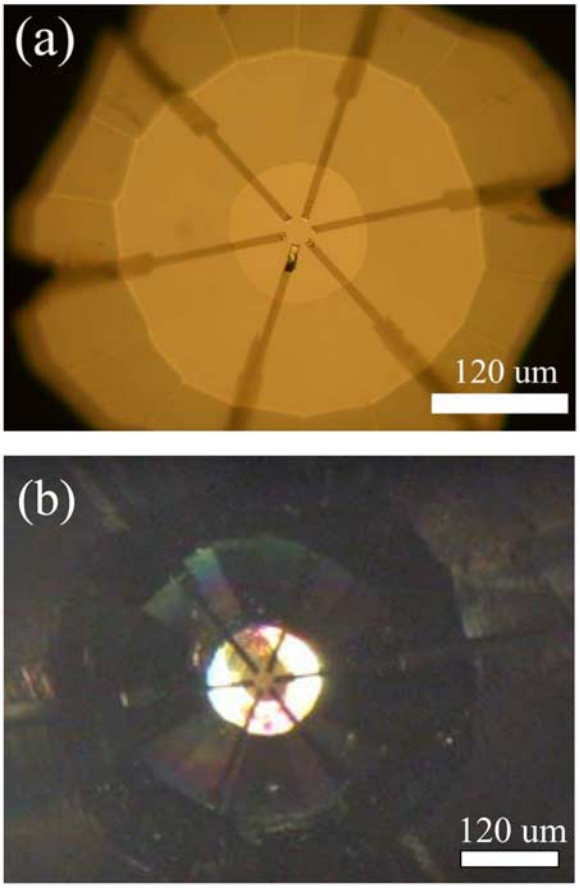

Figure 1. (a) Image of a designer diamond anvil with six tungsten probes used in the high-pressure electrical conductivity experiments (in both transmitted and reflected lights). The beveled anvil has an inner culet of 120 μm and an outer culet of 350 μm. The microprobes were made of thin-film tungsten $\sim$15 μm wide and $\sim$0.5 μm thick, and were covered by the epitaxially deposited diamond except small contact areas emerging from the synthetic diamond layer. (b) $(Mg_{0.75},Fe_{0.25})O$ sample in a DAC at 81 GPa (in reflected light). Re was used as the metal gasket to confine the sample of 60 μm in diameter. The sample at the central part of the inner culet was opaque to the transmitted light and showed dull reflection to the white light.

buffer. The sample was compositionally homogeneous in electron microprobe analyses and its ferric iron (Fe^{3+}) content was below the detection limit of Mössbauer spectroscopy, $Fe^{3+}/\Sigma Fe < 0.01$. X-ray diffraction of the sample showed that the sample was in the cubic rocksalt structure with an axial length (a) of 4.2411 Å ($\pm$0.0004). The powder sample was first compressed to $\sim$10 GPa in a separate DAC to squeeze out voids in between grain boundaries in the sample, and then loaded into the sample chamber of 60 μm in diameter in the designer DAC [*Weir et al.*, 2000] with two beveled diamonds with an inner culet of 80 to 120 μm and an outer culet of 350 μm (Figure 1). Tungsten electrical probes of the designer anvils were designed and fabricated at Lawrence Livermore National Laboratory, and the probes were then encased in epitaxially grown diamond layers at the University of Alabama at Birmingham [*Weir et al.*, 2000]. The resistance of each tungsten probe was approximately 100 Ω whereas the probe-to-probe leakage was more than 10 GΩ. Two sets of experiments with either a six-probe or four-probe designer anvil were conducted to over one megabar. In both cases, the electrical resistance was measured using a two-probe configuration by a Keithley 6517A electrometer in the DC current. Two-probe measurements are suitable for these experiments because the sample resistance is in the order of tens to hundreds of MΩ, much larger than the resistance of our measurement probes ($\sim$100 Ω). The use of the designer anvil also provided uniform and well-defined sample chamber geometry as the deformation of the sample and probe geometry can affect the resistance measurement in the DAC and complicate data interpretation [*Mao*, 1973; *Nellis et al.*, 1999]. We used the cell constant calculated in the three-dimensional current flow simulations [*Nellis et al.*, 1999] to relate the measured sample resistances to the electrical conductivities (Figure 2). Although our measured sample resistances are fairly accurate, conductivities obtained in this manner may be expected to have systematic errors of up to 30% [*Nellis et al.*, 1999], due mainly to the uncertainties in the sample geometry. Our low-pressure conductivities agreed to within 50% of previously reported electrical conductivities of ferropericlase [*Hansen and Cutler*, 1966; *Dobson and Brodholt*, 2000a].

[7] A small ruby sphere of 5 μm was also loaded into the sample chamber and used for pressure determination using the ruby fluorescence scale [*Mao et al.*, 1978]; pressures were also measured using the Raman frequency shift of the diamond first-order peak [*Occelli et al.*, 2003] at above

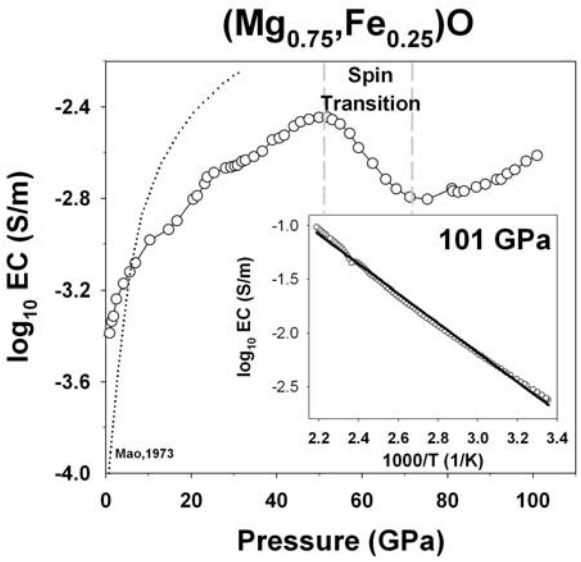

Figure 2. Electrical conductivities of $(Mg_{0.75},Fe_{0.25})O$ as a function of pressure obtained from a six-probe designer anvil cell. The conductivity increases by an order of magnitude up to 50 GPa but drops by a factor of approximately three from 50 to 70 GPa. Dotted line: electrical conductivities of $(Mg_{0.78},Fe_{0.22})O$ at high pressures [*Mao*, 1973]. Insert, electrical conductivities of $(Mg_{0.75},Fe_{0.25})O$ plotted against absolute reciprocal temperature at 101 GPa. Open circles: experimental data; solid line: fit to the Ahrrenius equation.

approximately 90 GPa when ruby fluorescence peaks became too weak to permit reliable pressure determination. High temperature measurements were conducted by heating the DAC with an external heater, and temperatures were measured from a K-type thermocouple attached to the diamond surface.

3. Experimental Results

[8] The electrical conductivity of $(Mg_{0.75},Fe_{0.25})O$ has been studied up to 101 GPa and 500 K (Figure 2) using a six-probe designer anvil. $(Mg_{0.75},Fe_{0.25})O$ remains a semiconductor in the pressure and temperature range investigated. The electrical conductivity of $(Mg_{0.75},Fe_{0.25})O$ increases with increasing pressure by an order of magnitude up to 50 GPa, though a separate study on $(Mg_{0.78},Fe_{0.22})O$ by *Mao* [1973] reported an increase of the conductivity by two orders of magnitude up to 32 GPa which was likely due to the relatively undefined sample and probe geometries. Further increase in pressure results in a decrease in the electrical conductivity of $(Mg_{0.75},Fe_{0.25})O$ by a factor of approximately three between 50 to 70 GPa (Figure 2). The electrical conductivity then increases slightly with increasing pressure above 70 GPa. Another set of measurements to 104 GPa using a four-probe designer anvil also showed a decrease in the electrical conductivity from 50 to 70 GPa by a factor of approximately two.

[9] The electrical conductivity of the low-spin ferropericlase has also been studied at high temperatures up to 500 K at 81 GPa and 101 GPa, respectively. The measured conductivities at high temperature were fitted to the Arrhenius equation to derive the activation energy, *Ea* (Figure 2, insert). The activation energy of the low-spin ferropericlase is 0.26 eV at 81 GPa and 0.27 eV at 101 GPa, consistent with the small polaron conduction (electronic hopping, charge transfer) but slightly lower than the *Ea* of 0.3 to 0.4 eV of the high-spin ferropericlase at relatively low temperatures [e.g., *Dobson et al.*, 1997; *Dobson and Brodholt*, 2000a].

4. Discussion and Geophysical Applications

[10] The observed decrease in the electrical conductivity between 50 to 70 GPa is consistent with the pressure range of the electronic spin-pairing transition in ferropericlase, where a mixture of the high-spin and low-spin states of ferropericlase has been recently reported to coexist across the transition [e.g., *Badro et al.*, 2003; *Lin et al.*, 2005, 2006a, 2006b, 2007; *Speziale et al.*, 2005; *Goncharov et al.*, 2006; *Persson et al.*, 2006; *Tsuchiya et al.*, 2006; *Keppler et al.*, 2007]. That is, the electronic spin transition of iron between 50 to 70 GPa results in a decrease in the mobility and/or density of the charge transfer carriers (small polaron) in the low-spin ferropericlase, which in turn decreases the electrical conductivity across the isosymmetric transition. Recent optical absorption spectra under high pressures showed that the spin transition of iron in ferropericlase at approximately 60 GPa enhances the optical absorption in the mid- and near-infrared spectral range, whereas the absorption edge at the ultraviolet region is reduced [*Goncharov et al.*, 2006; *Keppler et al.*, 2007]. The spin-pairing transition is estimated to reduce the radiative thermal conductivity of ferropericlase by about 15% [*Keppler et al.*, 2007] or more [*Goncharov et al.*, 2006] at ~60 GPa and room temperature. Our results here further show that the spin transition reduces the electrical conductivity of ferropericlase by a factor of approximately three.

[11] Field observations of the electrical conductivity profile of the Earth have shown that the electrical conductivity of the Earth's lower mantle is around one to tens S/m [*Olsen*, 1999], whereas a highly conducting layer with the conductance of $>10^8$ S may exist at the base of the lower mantle (the exact value of conductivity depends on the thickness of the layer) [*Holme*, 1998a, 1998b]. Because no measurements on the electrical conductivity of low-spin ferropericlase have been available up until now, laboratory measurements of the electrical conductivity of the high-spin ferropericlase [(Mg,Fe)O] and silicate perovskite [Al-(Mg,Fe)SiO_3] under high pressures and temperatures have been used to explain the field observations [*Xu et al.*, 2000]. Our results indicate that low-spin ferropericlase exhibits lower electrical conductivity than high-spin ferropericlase, which needs to be considered in future geomagnetic models for the lower mantle. High temperatures above ~1000 K may affect the conductivity mechanism in the low-spin state, because a large polaron conducting mechanism with an activation energy of 0.6 to 1 eV has been attributed to the high-temperature conductivity of the high-spin ferropericlase [*Roberts et al.*, 1995; *Dobson et al.*, 1997]. It remains to be seen how the theoretically predicted spin crossover of iron in ferropericlase that extends from the middle part to the lower part of the lower mantle [*Sturhahn et al.*, 2005; *Tsuchiya et al.*, 2006; *Lin et al.*, 2007] affect the electrical conductivity of ferropericlase as well as silicate perovskite at the lower mantle pressure-temperature conditions.

[12] Assuming that the structure and electrical conductivity mechanism of ferropericlase remain similar under lower mantle conditions [e.g., *Dubrovinski et al.*, 2000; *Lin et al.*, 2003; *Dobson et al.*, 1997], an extrapolation of the electrical conductivity of the low-spin ferropericlase to the lower-mantle pressure-temperature conditions (~2500 K and ~100 GPa) yields an electrical conductivity in the order of tens of S/m, consistent with the model values for the lower mantle [e.g., *Olsen*, 1999]. Such value is apparently too low to account for the possible existence of a highly conducting layer with the conductance of $>10^8$ S at the base of the lower mantle (the conductivity depends on the thickness of the layer) [*Holme*, 1998a, 1998b], though possible existence of metallic FeO and addition of ferrous iron can significantly alter the conductivity of the region [*Knittle and Jeanloz*, 1986; *Cohen et al.*, 1997; *Dobson and Brodholt*, 2005]. If the spin transition in ferropericlase results in the separation of two distinct phases with MgO-rich and FeO-rich compositions [e.g., *Dubrovinsky et al.*, 2000; *Lin et al.*, 2003], the occurrence of the metallic FeO could significantly affect current extrapolation of the electrical conductivity using semi-conducting ferropericlase to the lower mantle conditions. On the other hand, a highly conducting post-perovskite phase existing at the core-mantle boundary [*Ono et al.*, 2006] has been invoked to explain the exchange of the angular momentum and electromagnetic coupling between the liquid core and the solid mantle that result in the observed changes in the Earth's length of day [e.g., *Buffet*, 1992, 1996; *Holme*, 1998a, 1998b]. However,

the electrical conductivity of the post-perovskite phase is unknown and is only suggested to be much larger than the perovskite phase based on the analogous transition in Al_2O_3 [*Ono et al.*, 2006] that results in a two order of magnitude increase in the conductivity above 200 GPa [*Weir et al.*, 1996]. Future studies on the electrical conductivities of FeO and post-perovskite phase using our novel designer anvil technique with well fabricated electrical probes will help in understanding their potential roles in the geomagnetism at the core-mantle region.

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W. J. Evans, D. D. Jackson, J.-F. Lin, S. T. Weir, and C.-S. Yoo, Lawrence Livermore National Laboratory, 7000 East Avenue, Livermore, CA 94550, USA. (afu@llnl.gov)

W. Qiu and Y. K. Vohra, Department of Physics, University of Alabama at Birmingham, Birmingham, AL 35124, USA.