Unified understanding of the valence transition in the rare-earth monochalcogenides under pressure

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Valence instability is a key ingredient of the unusual properties of f electron materials, yet a clear understanding is lacking as it involves a complex interplay between f electrons and conduction states. Here we propose a unified picture of pressure-induced valence transition in Sm and Yb monochalcogenides, considered as a model system for mixed valent 4f-electron materials. Using high-resolution x-ray-absorption spectroscopy, we show that the valence transition is driven by the promotion of a 4f electron specifically into the lowest unoccupied (LU) $5d t_{2g}$ band. We demonstrate with a promotional model that the nature of the transition at low pressures is intimately related to the density of states of the LU band, while at high pressures it is governed by the hybridization strength.

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Valence fluctuations play an essential role in some of the f-electron systems' most exciting behaviors, such as quantum criticality and unconventional superconductivity.^{1,2} But their understanding has proven challenging to capture in a unified theory because they arise from subtle many-body interactions between the f electrons and the conduction states.^{3,4} Pressure, on the other hand, is an efficient way to act on f-electron interactions and localization and thus can serve as a window onto the f-electron physics. Notably, the pressure-induced valence transition of the lanthanide monochalcogenides is a paradigmatic illustration of f delocalization phenomena and their tremendous diversity. For instance, SmS undergoes a first-order transition to near trivalency coinciding with the onset of magnetic ordering,⁵ whereas the transition of YbS into an intermediate-valent correlated metal is extremely sluggish.⁶ These diverse behaviors, while establishing a severe test bed for theoretical understanding of f-electron systems, have been overlooked for the past decades. Here, we address them in the light of a direct measurement of the electronic structure of Sm and Yb monochalcogenides (SmS, SmSe, SmTe, YbS, YbSe) under pressure performed using high-resolution x-rayabsorption spectroscopy in the partial fluorescence yield (PFY-XAS) mode. Our data reveal that the valence discontinuity and concomitant closing of the gap under pressure are caused by the specific filling of the lowest unoccupied (LU) 5d t_{2g} band by a 4f electron, offering a unified picture of electron delocalization and intermediate valency. Using a promotional model, we show that both the steepness and the amplitude of the valence transition of the Sm compounds increase with the density of states (DOS) of this LU band. When exceeding a critical DOS, the transition becomes first order for SmS. The model fails to describe the prolonged transition of the Yb compounds, which suggests that at high pressures the valence transition is impeded by enhanced hybridization. Ultimately, we suggest that our analysis can serve as a prototype for understanding the general mechanism of delocalization in f-electron materials.

The PFY-XAS measurements were carried out at room temperature at the Taiwan beamline BL12XU, SPring-8. Details of the experimental setup have been published elsewhere.⁷ For each compound, a fragment of single crystal $\sim 100 \ \mu m$ in size was loaded into the sample chamber of a Be gasket with silicone oil used as pressure transmitting medium. Pressure was achieved using a diamond-anvil cell. The pressure dependence of the PFY-XAS spectra measured at the L_3 edge on the Sm and Yb monochalcogenides is shown in the top panels of Figs. 1(a) and 1(b). The spectra consist of two main peaks corresponding to the dipolar transition $2p \rightarrow 5d$ for the divalent and trivalent states, respectively decreasing and increasing under pressure. A closer look at the leading edge of the divalent peak reveals a shoulder located around 2.3 (SmS, YbS), 2.0 (SmSe, YbSe), and 1.6 (SmTe) eV below the peak maximum (open circles in Fig. 1). A fit of the divalent component actually requires the use of three Gaussian functions in order to reproduce the low-energy shoulder, the main peak, and a high-energy shoulder, hereafter referred to as A, B, and C, as shown in the center and bottom panels of Figs. 1(a) and 1(b).

In order to understand the origin of these spectral features, we have calculated the electronic structure of the Sm and Yb monochalcogenides using the density-functional theory within the generalized gradient approximation (GGA). The DOS obtained for the Sm 5d band and the crystal-field split

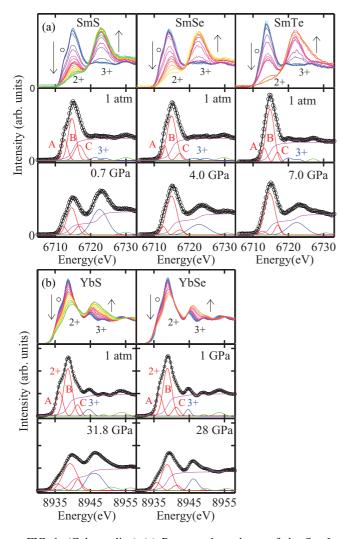


FIG. 1. (Color online) (a) Pressure dependence of the Sm- L_3 PFY-XAS spectra for SmS, SmSe, SmTe (top row), and examples of fit performed on spectra both before (center row) and after (bottom row) collapse. (b) Same as (a) for YbS and YbSe. For all five compounds, three components *A*, *B*, *C*, are used to fit the 2+ feature. Arrows indicate the direction of increasing pressure.

 t_{2g} and e_g subbands are compared with the experimental spectra in Fig. 2(a) for the Sm compounds. We note that the Coulomb interaction is not taken into account and therefore the calculated bands are metallic. Structures *A* and *C* are found to be respectively of pure t_{2g} and e_g character, while *B* is a mix of both. The energy splitting between the three structures and the relative intensity of *A* and *C* increase along the series SmTe < SmSe < SmS, which reflects the increase of the crystal field as the lattice constant decreases. We can therefore conclusively rule out the previous assignment of *A* to the quadrupolar pre-edge.^{5,8} Also, the band structures calculated for YbS in Fig. 2(b) and for the Sm compounds in Ref. 9 show that the low-energy part of *A*, which we call *A'*, corresponds to an isolated t_{2g} band sandwiched between the Fermi level and the rest of the conduction band, making it the LU band.

The pressure dependence of the valence v as derived from the fits of the PFY-XAS spectra is shown for the five monochalcogenides in Fig. 3(a). In previous estimations of v, A was incorrectly assigned to a quadrupolarlike feature

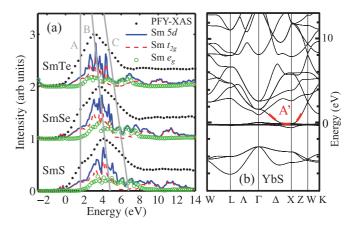


FIG. 2. (Color online) (a) Sm 5*d*, t_{2g} , e_g DOS calculations for SmS, SmSe, SmTe compared with the corresponding PFY-XAS spectra. As indicated by the solid lines, the energy shift between the peaks *A*, *B*, and *C* decreases across the series SmS, SmSe, SmTe. (b) Band structure of YbS. The low-energy, isolated part of the band *A*, *A'*, is highlighted.

and therefore not contributing to the divalent weight, which resulted in an overestimation of v by ~0.1 for SmS and 0.3 for YbS.^{5,8} All five compounds display a steady increase in v at their insulator-to-metal transition pressure, coinciding with the pressure ranges reported for their volume collapse.¹⁰ The abrupt increase in v around 15 GPa for SmTe is due to a structural transition. The onset pressure of the collapse gradually increases as a function of both the ligand (S < Se < Te) and more starkly the rare earth (Sm≪Yb). This reflects the difference in the gap values at ambient pressure, $\varepsilon_{gap} =$ 0.20 eV for SmS, 0.45 eV for SmSe, 0.65 eV for SmTe, 1.4 eV for YbS, and 1.75 eV for YbSe,^{11–14} and the idea that the wider the gap, the larger the pressure to close it.

We now address the large variety of slopes and amplitudes of the transitions, and its intimate connection with the LU band.

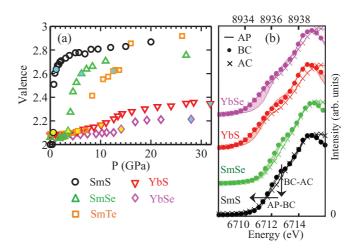


FIG. 3. (Color online) (a) Pressure dependence of the valence. Yellow and cyan filled markers respectively correspond to before collapse (BC) and after collapse (AC). (b) Enlarged divalent region of three spectra per compound normalized in intensity to the maximum of B, measured at ambient pressure (AP), BC, and AC. The arrows illustrate the low-energy shift of the shoulder A between AP and BC, and its collapse between BC and AC.

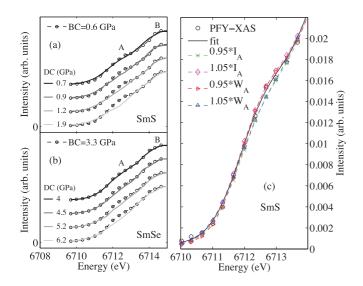


FIG. 4. (Color online) Evolution of feature A across the pressureinduced valence transition in SmS (a) and SmSe (b). For each compound, the Sm- L_3 PFY-XAS spectrum measured just before collapse (BC, dashed line and circles) is overlapped with four spectra measured during collapse (DC, solid line), all normalized in intensity to the maximum of B. (c) Illustration of the uncertainty on the fit around the region of shoulder A of SmS. Compared with the PFY-XAS spectrum (circles) are the fit (solid line), the fit where the as-derived value of I_A was decreased (dashed line and crosses) or increased (dashed line and diamonds) by 5%, and also the fit where the as-derived value of W_A was decreased (dashed line and right-pointing triangles) or increased (dashed line and up-pointing triangles) by 5%.

We start by showing in Fig. 3(b) the enlarged divalent region of three PFY-XAS spectra selected for four compounds, SmS, SmSe, YbS, YbSe, normalized in intensity to the maximum of *B*. These spectra were collected at ambient pressure (AP), just before the collapse (BC), and after the collapse (AC). The BC and AC pressures are respectively indicated by yellow and cyan filled markers in Fig. 3(a). The changes brought on by pressure on the line shape of A are significant. First, judging from the shape of the shaded area between the AP and BC spectra, A stretches towards low energies between AP and BC, and the extent of the stretch increases with the BC pressure P_{BC} , i.e., SmS < SmSe < YbS < YbSe. Second, the collapse coincides with a decrease of A, implying that A gets filled during the electronic transition. The decrease of shoulder A through the pressure-induced volume collapse is further illustrated in Fig. 4(a), where representative spectra measured at four successive pressures across the collapse are superimposed with the spectrum measured before the collapse for SmS and SmSe. All spectra are normalized in intensity to the maximum of the *B* feature.

In Fig. 5(b), the pressure dependence of key parameters derived from the fits of the PFY-XAS spectra gives a more precise view of the transition mechanism. For A, we use a phenomenological approach to subdivide the lower-energy portion of the peak which does not overlap with B, A', and the higher-energy portion which does, A'' [cf. Fig. 5(a)], in order to distinguish the filling of the low-energy part of the LU band truly isolated from the rest of the conduction band (A') from that of the higher-energy, hybridized part of the LU band

(A"). The sum $I_A + I_B + I_C$ is scaled to the total number of 5d holes, i.e., between 10 for v = 2 and 9 for v = 3. The pressure axis is rescaled by P_{BC} indicated by the vertical dashed lines. The error on the intensities and width is estimated to be under $\pm 5\%$, as illustrated in Fig. 4(c) where the effect on the fit of a $\pm 5\%$ error is shown.

Confirming the trend observed in Fig. 3, a close correspondence is observed between the increase of v and the decrease of A. More strikingly for Sm, the pressure ranges of the successive decline of $I_{A'}$ and $I_{A''}$ correlate respectively with the valence jump and the start of the slow increase [cf. blue circles and black crosses in Fig. 5(b)]. This demonstrates that the volume collapse transition results from the dumping of f electrons into A', while the transition notably slows down as soon as f electrons are promoted into A'', and slows even more when they start to fill the B band beyond AC. For YbS and YbSe the whole transition is considerably slower, as A' is not yet filled up at 30 GPa. The increase in I_{B+C} observed across the transition for all five compounds occurs as a result of B significantly broadening [cf. W_B in Fig. 5(b)], causing portions of A and C to merge with B. We attribute the concomitant decrease of A and C illustrated in Fig. 5(a) for SmS to increased merging of A and C with B. The amplitude of the valence discontinuity is therefore determined by the remaining intensity of A. Also noteworthy is the finding that the monochalcogenide gaps are closed by a combination of both band broadening and shift under pressure, as seen from the E_A and W_A panels.

For a systematic understanding, we have developed a simple model of the promotion of an electron from a 4*f* level of width zero into a square 5*d* band (corresponding to the LU band) of width *W* and occupancy *z*, as illustrated in Fig. 5(d). *W* is taken as 1 eV, which is an approximation of W_A at BC [cf. Fig. 5(b)], and *z* is taken as $v_{AC} - 2$. We consider that U_{fd} , the Coulomb interaction between the 4*f* and the conduction electrons, is effective. For $U_{fd} < \frac{W}{2z}$ the valence transition is continuous, and its pressure range, defined by the values of P_{BC} and P_{AC} , depends on the value of U_{fd} . In this case, the energy of the Fermi level ε_F is given by (cf. details in the Appendix)

$$\varepsilon_F = \frac{\varepsilon_f - 2U_{fd}\left(\frac{z}{W}\right)\left(E - \frac{W}{2}\right)}{1 - 2U_{fd}\left(\frac{z}{W}\right)},\tag{1}$$

where ε_f is the energy of the 4f level and E is the center of the square 5d band. The valence is then expressed as

$$v = 2 - \frac{z}{W} \left[\frac{E - \varepsilon_f - \frac{W}{2}}{1 - 2U_{fd} \left(\frac{z}{W}\right)} \right].$$
(2)

In our calculation of v, we put

$$E - \varepsilon_f - \frac{W}{2} = \varepsilon_{gap} - cP, \qquad (3)$$

with $c = \varepsilon_{gap} / P_{BC}$.

For $U_{fd} > \frac{W}{2z}$, at a critical value of the pressure there is a first-order transition and the valence changes discontinuously from 2 to 2 + z. The critical pressure is determined based on the condition that the total energy of the system under pressure is equivalent to the value at ambient pressure.

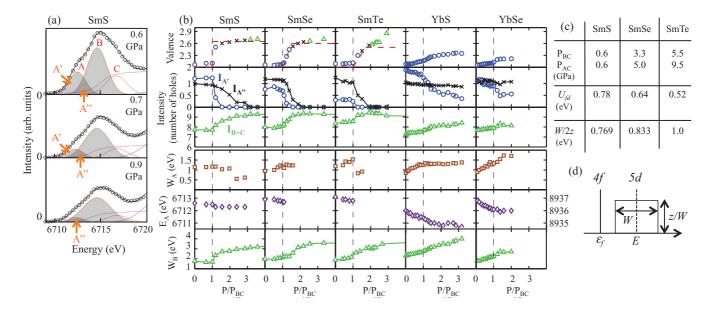


FIG. 5. (Color online) (a) Illustration of the decomposition of *A* into *A'* (light grey) and *A''* (dark grey). (b) Pressure dependence of *v* estimated experimentally (markers) and with the model (dashed line) (top row), and of the intensity of the peaks *A'*, *A''*, *B* + *C* (respectively second and third row from top). The use of different markers highlights the correspondence between the *v* transition and the filling of *A'*, *A''*, and *B* + *C* for the Sm compounds. Pressure dependence of the width and energy of the peak *A* (respectively third and second row from bottom), and of the width of the peak *B* (bottom row). The vertical dashed lines indicate BC pressure, P_{BC} , by which the pressure axis is rescaled. (c) Parameters of the promotional model. (d) Schematic illustration of the promotional model.

The transition for the Sm compounds is best reproduced for the parameters indicated in Fig. 5(c). Remarkably enough, the main features of the valence transition are properly captured for the three Sm compounds, a higher DOS of the LU band $\left(\frac{z}{W}\right)$ resulting in a steeper transition, ultimately first order for SmS, as shown in the top panel of Fig. 5(b). This lends support to the relevance of the promotional model, which has been an open question since the work of Ramirez and Falicov on Ce.¹⁵ Because our model describes the linear filling of one band, it approximates a constant slope through the transition between BC and AC, and does not account for the filling of A" and B beyond AC. For YbS and YbSe, our model yields unphysical negative values of U_{fd} . For YbS for instance, using in Eqs. (2) and (3) the experimental values $P_{BC} = 8.7 \text{ GPa}$, $P_{AC} = 18.0 \text{ GPa}$, z = 0.4, and $\varepsilon_{gap} = 1.4$ eV, together with W = 1.0 eV, we obtain $U_{fd} = -0.62$ eV. A very plausible explanation is that hybridization effects between the 4f electron and the 5d band become stronger at the higher transition pressures of the Yb compounds, so that A' no longer retains its isolated character.

This analysis thus delineates two broad scenarios for the electronic transition of the monochalcogenides. In the first, the LU band is isolated from the rest of the conduction band, allowing for a steep transition, whose slope and amplitude are proportional to the DOS of the LU band. Hybridization thrives in the second scenario, which leads to a sluggish transition, for which a Kondo description may be appropriate. We note that because $\frac{z}{W}$ decreases and hybridization increases with increasing pressures. This approximation holds for other rare-earth monochalcogenides, the transition being first order for TmSe_{0.45}Te_{0.55} at 1 GPa,¹⁶ second order but steep for TmTe at 2 GPa,¹⁷ and sluggish between 16 and 20 GPa for EuSe.¹⁸ Our interpretation may also be generalized to other

divalent 4*f* materials. EuO, for instance has an *A* shoulder in its L_3 spectrum, and a sluggish transition was recently reported above 14 GPa.¹⁹ On the other hand, divalent 4*f* materials without an *A* shoulder show a sluggish transition which starts as soon as above 1 atm, see, for instance, Pr,²⁰ Eu,²¹ Yb,²² YbAl₂,²³ YbGa_{1.15}Si_{0.85},⁷ or Cd_{5.7}Yb.²⁴ We believe this is because *f* electrons are directly promoted into the main 5*d* conduction band where hybridization effects are strong, a simple promotional model is therefore no longer sufficient. This is also likely the case for intermediate valent and trivalent compounds as they already show a strong hybridization at ambient pressure.

We have presented a simple framework which allows for a unified description of the diverse pressure-induced electronic phase transitions of Sm and Yb monochalcogenides, with implications for the 4*f* delocalization phenomenon in general. We have shown that the transition corresponds to the filling of the LU band. This band is at the heart of the diverse electronic transitions of the monochalcogenides, as the slope and amplitude of the transition depend on both its DOS and degree of hybridization. Our analysis can also seemingly describe the delocalization in a variety of other compressed 4*f* electron materials. It may also be extended to the 5*f* electron systems, which have a broad 6*d* unoccupied band and usually show protracted electronic transitions under pressure.²⁵

APPENDIX

We first explain how Eq. (1) is derived. If we assume that $n \ 4f$ electrons per Sm are promoted to the 5d band in the system sketched in Fig. 5(d), the value of n, and therefore of the Fermi level ε_F , is determined for the stable state by minimizing the total energy change $E_{band} + E_{4f} + E_{int}$ with

respect to *n* (or ε_F):

$$\frac{d}{d\varepsilon_F}(E_{band} + E_{4f} + E_{int}) = 0, \tag{A1}$$

where E_{band} and E_{4f} are respectively the kinetic energy changes of the electrons in the 5*d* band and of the 4*f* electrons per Sm, and E_{int} the Coulomb interaction energy between the 5*d* band electrons and the 4*f* holes per Sm:

$$E_{band} = \frac{z}{2W} \left[\varepsilon_F^2 - \left(E - \frac{W}{2} \right)^2 \right], \tag{A2}$$

$$E_{4f} = \left[1 - \frac{z}{W} \left(\varepsilon_F - E + \frac{W}{2}\right)\right] \varepsilon_f, \qquad (A3)$$

$$E_{int} = -U_{fd} \left[\frac{z}{W} \left(\varepsilon_F - E + \frac{W}{2} \right) \right]^2, \qquad (A4)$$

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with W and z being respectively the width and the occupancy in the postcollapse state of the 5d band, and U_{fd} the Coulomb interaction between the 4f and the conduction electrons.

Next, the expression for the valence in Eq. (2) of the paper is obtained by v = 2 + n, with ε_F and *n* being connected by the following relation:

$$n = \frac{z}{W} \bigg[\varepsilon_F - E + \frac{W}{2} \bigg]. \tag{A5}$$

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